1.4 Linear First-order Equations

Example. A reservoir holds 160,000 acre ft of water, 0.25% of which is a pollutant. Each day, 10,000 acre ft of water with pollutant concentration 0.05% enters the reservoir, and the same volume of well-mixed water leaves.

- (a) Set up a differential equation for the volume V(t) of pollutant in the reservoir after t days.
- (b) How long does it take for the pollutatant level to fall to 0.1%?

Solution. (a) Set up notation. Over a time interval Δt , we have an increase in pollutant volume ΔV . It should be clear that

$$\Delta V \approx (V_{\rm in} - V_{\rm out}) \Delta t$$

where V_{in} and V_{out} are the volumes of pollutant entering and leaving the reservoir *each day*. Now,

 $V_{\rm in} = ({
m Volume~of~water~entering~each~day}) \cdot ({
m Concentration}) = 10,000 \cdot rac{0.05}{100} = 5~{
m acre~ft},$ $V_{
m out} = ({
m Fraction~of~reservoir~leaving~each~day}) \cdot ({
m Volume~of~pollutant~in~reservoir})$ $= 10,000 \cdot rac{V(t)}{160,000} = rac{1}{16} V ~{
m acre~ft}.$

Therefore

$$\frac{\Delta V}{\Delta t} \approx 5 - \frac{1}{16}V.$$

Taking the limit as $\Delta t \rightarrow 0$ yields the differential equation

$$\frac{\mathrm{d}V}{\mathrm{d}t} = 5 - \frac{1}{16}V.$$

(b) The equation is 1st order linear: $V' + \frac{1}{16}V = 5$, so we use the integrating factor method:

$$r(t) = \exp\left(\int \frac{1}{16} \, \mathrm{d}t\right) = e^{t/16},$$

whence²

$$V(t) = \frac{1}{r(t)} \left[\int 5r(t) dt + A \right] = e^{-t/16} \left[\int 5e^{t/16} dt + A \right]$$

= 80 + Ae^{-t/16} acre ft.

Now consider the initial condition: if C(t) is the concentration, then

$$C(0) = 0.25\% \Rightarrow V(0) = 160,000 \cdot \frac{0.25}{100} = 400 \Rightarrow A = 320.$$

¹Note the variables: V and t, not y and x!

²Using the arbitrary constant A to avoid confusion with *concentration*.

Thus

$$V(t) = 80(1 + 4e^{-t/16})$$
 acre ft.

$$C(t) = 0.1\%$$
 when $V(t) = 160$, which is when

$$4e^{-t/16} = 1 \Rightarrow t = 16 \ln 4 \approx 22.18 \text{ days.}$$

Note: We could instead have gone for a differential equation for $\mathcal{C}(t)$ directly

$$\frac{dC}{dt} + \frac{1}{16}C = \frac{1}{32,000}$$

but concentration, being unitless, is a little harder to work with.

